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2	CAMx Ozone Source Attribution in the Eastern United States using Guidance from
3	Observations during DISCOVER-AQ Maryland
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# 25 **Key points**

- Ozone is simulated well by CAMx, but there is an overestimate of NO<sub>y</sub> and underestimate of HCHO.
- Ozone production in the new model framework is more sensitive to NO<sub>x</sub>
  emissions.
- Point sources likely contribute more to surface ozone than commonly appreciated.

# 32 Abstract

33	A Comprehensive Air-Quality Model with Extensions (CAMx) version 6.10 simulation
34	was assessed through comparison with data acquired during NASA's 2011 DISCOVER-
35	AQ Maryland field campaign. Comparisons for the baseline simulation (CB05
36	chemistry, EPA 2011 National Emissions Inventory) show a model overestimate of $NO_y$
37	by +86.2% and an underestimate of HCHO by -28.3%. We present a new model
38	framework (CB6r2 chemistry, MEGAN v2.1 biogenic emissions, 50% reduction in
39	mobile NO <sub>x</sub> , enhanced representation of isoprene nitrates) that better matches
40	observations. The new model framework attributes 31.4% more surface ozone in
41	Maryland to electric generating units (EGUs) and 34.6% less ozone to on-road mobile
42	sources. Surface ozone becomes more NO <sub>x</sub> -limited throughout the eastern United States
43	compared to the baseline simulation. The baseline model therefore likely underestimates
44	the effectiveness of anthropogenic $NO_x$ reductions as well as the current contribution of
45	EGUs to surface ozone.

## 1. Introduction

- 47 Policymakers and regulatory agencies use regional air quality models to predict how
- future air quality will respond to control strategies [EPA, 2014a]. Many air quality
- 49 models can skillfully simulate surface ozone in North America for focused studies of
- certain time periods [Hogrefe et al., 2004; Appel et al., 2007; Ferreira et al., 2011; Appel
- et al., 2012; Simon et al., 2012]. Global models can reflect changes in ozone resulting
- from control measures [e.g., Clifton et al., 2014; Rieder et al., 2015], especially for rural
- sites representative of regional atmospheric composition, but nonattainment is based on
- monitors with the highest readings. Urban-scale events, such as seen in Edgewood, MD,
- discussed below, require urban scale resolution of 12 km or better [e.g., Loughner et al.,
- 56 2011; Goldberg et al., 2014].
- 57 Even where regional air quality models accurately reproduce surface ozone
- concentrations, many have difficulty simulating the response of ozone to reductions in
- 59 precursor emissions [Gilliland et al., 2008; Zhou et al., 2013; Foley et al., 2015]. This
- may be linked to the challenge of simulating ozone precursors:  $NO_x$  ( $NO_x = NO + NO_2$ )
- and volatile organic compounds (VOCs) [Castellanos et al., 2011; Zhou et al., 2013;
- 62 Canty et al., 2015]. For any given ozone concentration, there can be many different
- production pathways; empirical kinetic modeling approach (EKMA) diagrams [Kinosian,
- 64 1982; Chameides et al., 1992; Sillman, 1999], highlight this non-linear dependence of
- ozone production on NO<sub>x</sub> and VOCs. Air quality models must be in the correct ozone
- production regime (i.e., NO<sub>x</sub>-limited vs. VOC-limited) if they are to accurately forecast
- 67 how air quality regulations will improve ozone concentrations.
- Many studies show an overestimate, by up to a factor of two, of total reactive oxidized
- 69 nitrogen (NO<sub>v</sub>) in regional air quality models compared to observations [Doraiswamy et
- 70 al., 2009; Castellanos et al., 2011; Yu et al., 2012; Brioude et al., 2013; Anderson et al.,
- 71 2014; Goldberg et al., 2014]. Some link the calculation of too much NO<sub>v</sub> to the
- overestimate of NO<sub>x</sub> emissions from area sources [Doraiswamy et al., 2009], while others
- link it to an overestimate of NO<sub>x</sub> emissions from commercial marine vessels [Brioude et
- al., 2013]. Anderson et al. [2014] examined airborne observations of CO, NO<sub>x</sub>, and NO<sub>y</sub>
- obtained in the Baltimore-Washington corridor and concluded that a substantial portion
- of the error must be due to an overestimate in NO<sub>x</sub> emissions from mobile sources since
- this source accounts for the majority (62%) of NO<sub>x</sub> emissions in the 2011 National
- 78 Emissions Inventory (NEI). Fujita et al. [2012] also find an overestimate of NO<sub>x</sub> mobile
- 79 source emissions in MOVES 2010a, which is used to develop the NEI.
- A better representation of NO<sub>v</sub> chemistry may resolve a portion of the overestimate of
- NO<sub>v</sub> noted above. The Carbon Bond 6 Revision 2 (CB6r2) gas-phase chemistry has been
- released recently [Hildebrandt-Ruiz and Yarwood, 2013]. This updated mechanism more
- 83 explicitly represents alkyl nitrates in regional air quality models and provides a

- significant improvement in the simulation of these compounds compared to CB05
- 85 [Hildebrandt-Ruiz and Yarwood, 2013; Canty et al., 2015]. CB6r2 splits the alkyl nitrate
- grouping (NTR) into three families: alkyl nitrates that exist primarily in the gas phase
- 87 (NTR1), larger multi-functional alkyl nitrates that partition to organic aerosol (NTR2)
- and isoprene nitrates (INTR) that react rapidly with OH. NTR1 and INTR can recycle
- back to NO<sub>2</sub>, but the only gas-phase sink for NTR2 is conversion to HNO<sub>3</sub>. The CB6r2
- 90 gas-phase mechanism calculates a shorter lifetime of alkyl nitrates and faster recycling of
- 91 NO<sub>x</sub>, which agrees better with laboratory studies [Perring et al., 2013] than CB05. In
- addition to improving the representation of alkyl nitrates in the regional air quality
- 93 models, this change may also improve the simulation of ozone attributed to sources
- beyond state borders. To further improve the representation of alkyl nitrates in air quality
- 95 models, Horowitz et al. [2007] suggest increasing isoprene nitrate deposition velocities.
- As anthropogenic sources of ozone precursors continue to decrease, biogenic emissions
- 97 will play an even larger role in the ozone formation process. Two models are used to
- 98 simulate biogenic emissions within regional air quality models: Biogenic Emissions
- 99 Inventory System (BEIS) [Pouliot and Pierce, 2009] and Model of Emissions of Gases
- and Aerosols from Nature (MEGAN) [Guenther et al., 2012]. Isoprene emissions are
- uniformly larger in the MEGAN model within North America than in BEIS [Warneke et
- 102 al., 2010; Carlton and Baker 2011].

## 2. Methods

- We use the Comprehensive Air-quality Model with Extensions (CAMx) version 6.10 to
- simulate trace gas mixing ratios in the eastern United States for July 2011; the model
- domain is shown in Figure S1. Many previous studies have used CAMx to simulate
- ozone with reasonable fidelity [Emery et al., 2012; Dolwick et al., 2015; Koo et al.,
- 108 2015]. The Anthropogenic Precursor Culpability Assessment (APCA) probing tool in
- 109 CAMx is used as a means to tag ozone source attribution from twelve source regions and
- seven source sectors. The twelve source regions are shown in Figure S2. The seven
- source sectors are listed in Table S1. We also use the Ozone Source Apportionment Tool
- 112 (OSAT) to calculate the ozone attributed to NO<sub>x</sub>- and VOC-limited production regimes.
- For a detailed description of CAMx v6.10 and the APCA and OSAT probing tools, please
- refer to the CAMx User's Guide [Ramboll Environ, 2014]. CAMx was driven off-line by
- meteorological output [EPA, 2014b] from the WRF v3.4 model [Skamarock et al., 2008]
- at hourly intervals. Specific details about the meteorology simulation are in the EPA
- technical support document [EPA, 2014b]. Table S2 describes the CAMx options chosen
- for our baseline simulation.
- We use version 2 of the 2011 NEI as compiled by EPA for anthropogenic emissions in
- our baseline simulation [EPA, 2014c]. The Continuous Emission Monitoring System
- 121 (CEMS) database temporalized by Eastern Regional Technical Advisory Committee

- 122 (ERTAC) software was used to create electric generation unit (EGU) emissions. This
- inventory allocates larger emissions of NO<sub>x</sub> during hotter days due to increased electricity
- demand [He et al., 2013], but does not include an estimate of additional NO<sub>x</sub> emitted by
- small peaking units. Mobile emissions estimates from cars, trucks, and motorcycles were
- computed with the Motor Vehicle Emission Simulator 2014 (MOVES2014) [EPA,
- 2014c]. Biogenic emissions in the baseline simulation were calculated using BEIS
- version 3.6 [Pouliot and Pierce, 2009]. The Mid-Atlantic Regional Air Management
- 129 Association (MARAMA) prepared total emissions for our model domain. Boundary
- conditions were initialized using the GEOS-Chem v8-03-02 global chemistry model [Bey
- et al., 2001] at a horizontal resolution of 2.0° latitude × 2.5° longitude, as described in
- 132 Henderson et al. [2014].

#### **3. Results**

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#### 3.1 Baseline Model Simulation

- During July 2011, NASA conducted a comprehensive aircraft and ground measurement
- campaign in Maryland called DISCOVER-AQ. This campaign provided a temporally-
- and spatially-rich collection of trace gas and aerosol observations throughout the lower
- troposphere [Crawford et al., 2014]. This dataset offers an unprecedented opportunity to
- compare regional air quality models to comprehensive atmospheric observations.
- Figure 1 (left) compares ozone (O<sub>3</sub>) and two important ozone precursors, NO<sub>v</sub> and
- 141 formaldehyde (HCHO), from the baseline model simulation to P3-B aircraft observations.
- All observations were taken between altitudes of 300 5000 m within the Maryland
- airshed. In the top left the scatterplot of modeled ozone vs. observed ozone we show
- a slope near unity (1.06) and a normalized mean bias (NMB) of -6.90% indicating a
- small underestimate of ozone above the surface. Because the NMB is under 10%, the
- baseline simulation shows good agreement with the observations of ozone. The root-
- mean square error (RMSE) of the baseline simulation of ozone is 9.88 ppbv. In the
- supplementary material, Figure S3, we provide a comparison to surface observations,
- which shows even better agreement with the baseline simulation.
- 150 Comparing modeled NO<sub>v</sub> and HCHO to observations of the same quantities shows large
- discrepancies. The model simulation overestimates NO<sub>v</sub> by nearly a factor of two: a
- slope of 1.91 and a NMB of +86.2%. An overestimate of NO<sub>v</sub> is also seen at the
- Edgewood, Maryland ground site as shown in Figure S4; instrument description is
- provided in Martins et al. [2012]. Conversely, the model simulation underestimates
- HCHO by nearly a factor of two: a slope of 0.61 and a NMB of -28.3%. Although ozone
- is being predicted with considerable skill, the ozone precursors (NO<sub>v</sub> and HCHO) are not.
- In the supplementary material, Figures S5, S6, S7 and S8, we show comparisons of NO<sub>2</sub>,
- alkyl nitrates, nitric acid, and isoprene.

- The overestimate of NO<sub>v</sub> and underestimate of HCHO by the baseline model simulation
- are more pronounced at the lowest altitudes of the P3-B aircraft spirals. In Figure 2, we
- show vertical profiles of measured ozone, NO<sub>v</sub>, and HCHO binned in 500 m intervals and
- the closest CAMx model grid point, matched spatially and temporally during all flights.
- 163 The median value of observed NO<sub>y</sub> at the lowest altitude is below the 25<sup>th</sup> percentile of
- simulated NO<sub>y</sub>, while the median value of observed HCHO is above the 75<sup>th</sup> percentile of
- simulated HCHO. We also find that ozone is underestimated for the lowest sampled
- altitudes, but agrees well with observations above 2.5 km; the underestimate of ozone,
- however, is not seen directly at the surface (Figure S3).

## 3.2 Updated "Beta" Model Simulation

- We update the CAMx model platform based on recommendations from recent scientific
- 170 literature outlined in the Introduction. The four changes are:
- Update the gas-phase chemistry from CB05 to CB6r2, which better
- represents alkyl nitrate photochemistry [Hildebrandt-Ruiz and Yarwood,
- 173 2013].

- Update the biogenic emissions from BEIS v3.6 to MEGAN v2.1, which
- increases isoprene emissions [Guenther et al., 2012].
- Reduce NO<sub>x</sub> emissions from mobile sources (on-road, off-road and non-
- road) by 50% within our model domain [Anderson et al., 2014].
- Increase the dry deposition velocities of isoprene nitrates (INTR) and
- multi-functional alkyl nitrates (NTR2) to be the same as nitric acid
- 180 (HNO<sub>3</sub>) [Horowitz et al., 2007].
- We label the CAMx simulation with these four changes as the "Beta" simulation and
- compare the same trace gases (O<sub>3</sub>, NO<sub>v</sub>, HCHO) from this updated run to P3-B aircraft
- observations in the right side of Figure 1. The Beta simulation exhibits substantial
- improvement in the estimate of ozone precursors. The NMB of NO<sub>v</sub> has improved from
- +86.2% to +22.4% and the NMB of HCHO has improved from -28.3% to -0.47%. The
- 186 RMSE of NO<sub>v</sub> and HCHO both improve: NO<sub>v</sub> from 3.09 ppbv to 1.71 ppbv and HCHO
- 187 from 1.34 ppbv to 0.93 ppbv. The NMB of NO<sub>v</sub> at the Edgewood, MD ground monitor
- also improves from +46.9% to -7.8% using this new model platform (Figure S4). The
- Beta simulation yields similar predictions of ozone compared to the original calculation:
- the baseline has a NMB of -6.90%, whereas the Beta simulation has a NMB of -7.82%.
- The RMSE of the ozone degrades slightly from 9.88 ppbv to 10.53 ppbv. Deteriorating
- 192 performance of ozone in the Beta simulation may be due to not enough recycling of
- multi-functional alkyl nitrates to NO<sub>2</sub> in the CB6r2 gas-phase mechanism.
- The Beta simulation also shows better agreement with the vertical profiles of NO<sub>v</sub> and
- 195 HCHO (Figure 2). The median value of observed NO<sub>v</sub> is much closer to the median

- value of modeled NO<sub>y</sub>. At altitudes above 2.5 km, there is no improvement in the
- simulation of NO<sub>y</sub>, likely due to an overestimate of HNO<sub>3</sub> within the GEOS-Chem global
- model used to initialize the CAMx boundaries (Figure S9). At these altitudes, HNO<sub>3</sub> is
- 199 photochemically inactive and the overestimate will have minimal impact on ozone
- formation. The median value of observed HCHO is also much closer to the median value
- of HCHO from the Beta simulation. However, there is now a large overestimate in the
- simulation of isoprene (Figure S6), which suggests errors in the isoprene to formaldehyde
- 203 conversion processes in CB6r2. Mao et al. [2013] show that improvements to isoprene
- 204 oxidation processes in air quality models are still needed. We also compare the isoprene
- observations to a CAMx simulation with a recently released version of BEIS v3.61 [Bash
- et al., 2015], which shows the best agreement with observations (Figure S10); BEIS
- v3.61 has improved land-use and canopy representation. Similar to our study, Kota et al.
- 208 [2015] also showed an overestimate of isoprene using MEGAN v2.1 in southeast Texas.
- The comparison of observed ozone to values from the Beta simulation exhibits similar
- 210 features as the comparison for the baseline simulation. The NMB of seven trace gases for
- 211 the baseline, each modification isolated separately, and Beta simulations are given in
- Table S3.

## 3.3 Changes to Ozone Attributed to Mobile vs. Large Point Sources

- The NEI shows on-road and off-road mobile source emissions account for the largest
- portion of the total NO<sub>x</sub> emissions, 61% of the total (Figure S11). In Maryland the
- 216 percentage is even larger; NO<sub>x</sub> emissions from on-road and off-road sources account for
- 217 72% of total NO<sub>x</sub> emissions. Figure 3 depicts ozone attributed to emissions from
- 218 individual states (denoted by color) as well as from various source sectors (each
- 219 histogram). Results are shown for both the (left) baseline and (right) Beta simulations,
- for the ten worst modeled air quality days in July 2011 at Edgewood, Maryland; observed
- surface ozone during these ten days is 81.3 ppbv (only six of the top ten worst modeled
- days match the top ten worst observed days). We have chosen to focus on Edgewood
- 223 (the location shown as the filled circle in Figure 4) because this site causes the Baltimore
- region to be in moderate non-attainment of the 2008 NAAQS for ozone [EPA, 2014a]. In
- 225 the baseline simulation (Figure 3, left) generated from the NEI on-road sources are
- responsible for the largest portion (24.6 ppbv) of total surface ozone. Ozone attributed to
- 227 electric generating units (EGUs) accounts for the second largest single sector (11.6 ppbv)
- during the ten worst air quality days at Edgewood. The NEI indicates EGUs are
- responsible for 14% of total NO<sub>x</sub> emissions, and 11% within the state of Maryland.
- 230 In the Beta simulation we keep emissions from EGUs identical to the baseline simulation
- because the NEI is developed from observed Continuous Emissions Monitoring System
- 232 (CEMS) data. There is strong scientific basis [Anderson et al., 2014] to link the
- overestimate in NO<sub>v</sub> to mobile source emissions since they represent more than 50% of
- 234 the NO<sub>x</sub> emissions inventory. The Beta simulation (Figure 3, right) attributes more ozone

- 235 to EGUs and less ozone to mobile sources. While on-road mobile sources are still the
- primary individual source sector contributing to surface ozone, they are responsible for
- 7.7 ppbv less ozone compared to the baseline simulation: 24.6 ppbv to 16.9 ppbv, a drop
- of 31.4%. Ozone attributed to non-road sources also shows a similar percentage drop.
- 239 Despite identical emissions of NO<sub>x</sub> from EGUs in the two simulations, electricity
- 240 generation is responsible for 4.0 ppbv more ozone in the Beta run, increasing from 11.6
- to 15.6 ppbv, a 34.6% increase. The ozone attributed to EGU emissions shows a large
- increase because CB6r2 gas-phase chemistry has faster photolysis of NO<sub>2</sub> than CB05 and
- increased modeled HO<sub>2</sub> and RO<sub>2</sub> concentrations driven by greater biogenic emissions
- from MEGAN v2.1. This implies greater ozone production efficiency, a topic to be
- treated in a separate paper. For the Beta simulation, EGUs and on-road mobile sources
- are now responsible for roughly the same fraction of surface ozone in Maryland. The
- 247 change in surface ozone attribution to on-road mobile and EGU sources for the baseline
- compared to the Beta simulation is similar throughout the eastern United States for July
- 249 2011 (Figure S12).

## 3.4 Changes to Ozone Attributed to NO<sub>x</sub> & VOC limitations

- The overestimate of NO<sub>v</sub> and underestimate of HCHO in the baseline simulation suggests
- 252 that ozone in the original model framework may be produced in a more VOC-limited
- 253 ozone production regime than occurs in the actual atmosphere, even though NO<sub>x</sub> remains
- 254 the key pollutant. To better grasp the relationship between modeled and observed ozone
- precursors, we plot ozone as a function of NO<sub>v</sub> for the observations and two model
- simulations (Figure S13). The observed slope of the linear-best fit indicates 20.9 ppbv of
- ozone per ppbv of NO<sub>v</sub> in the Maryland airshed, whereas, the baseline simulation
- 258 indicates a slope of 8.6. Ozone becomes more sensitive to NO<sub>v</sub> in the updated Beta
- 259 model platform, which yields a slope of 13.3. We also compare HCHO as a function of
- NO<sub>v</sub> (Figure S14). The linear best fit of the observations show 1.39 ppbv of HCHO per
- ppbv of NO<sub>v</sub>; the baseline model has a linear fit of 0.45, but the Beta simulation show a
- slope of 1.28, which is closer to the observations. The sensitivity of ozone to the
- abundance of its precursors is captured better in the updated Beta model platform.
- We also use an OSAT simulation to calculate the amount of ozone formed in NO<sub>x</sub>-limited
- and VOC-limited environmental conditions. Figure 4 shows the percentage of ozone
- production attributed to a NO<sub>x</sub>-limited ozone regime. In the baseline simulation, 65 –
- 267 85% of ozone in the Baltimore vicinity is attributed to a NO<sub>x</sub>-limited environment. The
- 268 updated Beta simulation uniformly shows more ozone production in a NO<sub>x</sub>-limited
- regime. The biggest differences occur over the Chesapeake Bay. The Beta simulation
- shows 80 95% of ozone is produced in a NO<sub>x</sub>-limited environment in the Baltimore
- 271 vicinity. Instead of being in the "transition region" the region on the EKMA diagram in
- which ozone production occurs due to both VOC and NO<sub>x</sub> limitation the area is now

- 273 squarely in a region of NO<sub>x</sub>-limited ozone production. This is consistent with observed
- 274 changes in ozone resulting from NO<sub>x</sub> emission reductions [Gilliland et al., 2008].

## 3.5 Changes to Ozone Source Region Attribution

- 276 Modifications to the model framework do not have a big effect on source attribution, but
- subtle differences are worth discussing. Figure S15 shows state-by state attribution at the
- 278 Edgewood, Maryland monitor for the ten worst modeled air quality days during July
- 279 2011 for the baseline and Beta simulations. Maryland is the largest contributor to total
- ozone mixing ratios at Edgewood. States upwind of Maryland during hot summertime
- days, i.e. Pennsylvania, Virginia, and Ohio contribute more than 4 ppbv each. Further
- discussion on the interstate transport of ozone is included in Goldberg et al. [2015].
- 283 When changing model platforms, Maryland shows a slight rise in attribution (27.9 ppbv
- to 29.1 ppbv), while other states show small declines in ozone attribution (i.e., Virginia).
- 285 The changes do not shift any state from being above or below 1 ppbv a critical value
- legislated by the Cross-State Air Pollution Rule (CSAPR).
- Each individual incremental change to the modeling platform alters the source region
- attribution. Figure S16 shows source region attribution of surface ozone at Edgewood
- during the ten worst air quality days in July for five simulations in three scenarios:
- baseline, baseline with CB6r2 and increased alkyl nitrate depoistion, baseline with
- 291 MEGAN v2.1 biogenics, baseline with 50% mobile NO<sub>x</sub> emissions, and Beta. For the
- baseline simulation (left), Maryland is responsible for 30.9% of the total; interstate
- transport accounts for the other 69.1%. Improvement of the alkyl nitrate photochemistry
- and the mobile emissions inventory make ozone photochemistry more of a regional
- problem, as shown by the slightly reduced contributions from Maryland in the
- 296 CB6r2+NTRdepn and 50% mobile NOx simulations, 29.3% and 30.0%
- 297 respectively. Changes to the biogenic emissions inventory, resulting in increased
- isoprene, make ozone photochemistry more of a local issue, with Maryland's contribution
- in the MEGAN v2.1 increasing to 36.0%.

#### 4. Conclusion

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- 301 CAMx, when modified with guidance provided by a field experiment, more realistically
- 302 simulates the observed abundance of ozone precursors. We compare ozone precursors
- 303 (NO<sub>v</sub> and HCHO) and ozone measured during the July 2011 DISCOVER-AQ Maryland
- campaign to CAMx simulations. In the baseline simulation, there is good agreement
- between modeled and observed ozone, but poor agreement for NO<sub>v</sub> and HCHO. We
- implemented four changes to the model: CB6r2 gas-phase chemistry, faster deposition of
- alkyl nitrates, reduced NO<sub>x</sub> emissions from mobile sources, and increased isoprene
- 308 emissions by switching to MEGAN v2.1 biogenic emissions. Our results indicate that
- 309 BEIS v3.61 shows good agreement with isoprene observations, and we recommend this

310	over BEIS v3.6. The Beta runs dramatically improve the simulation of total reactive
311	nitrogen, alkyl nitrates, and formaldehyde. Adding more recycling of alkyl nitrates to
312	NO <sub>2</sub> in CB6r2 and refining isoprene photochemistry may further improve CAMx
313	performance.
314	These modifications change the attribution of ozone to different source sectors and have
315	important policy implications. Compared to the baseline simulations, mobile sources
316	contribute 31.4% less to total ozone while EGUs contribute 34.6% more at Edgewood,
317	Maryland. Ozone attributed to EGUs increase from 11.6 to 15.6 ppbv, while ozone
318	attributed to mobile sources decreases from 24.6 to 16.9 ppbv. Ozone in the two model
319	simulations is comparable and agrees reasonably well with observations, but the source
320	attribution and targets for control strategies change substantially.
321	Prior research demonstrated that regional air quality models underestimate the benefit of
322	NO <sub>x</sub> control measures for surface ozone. If air quality models are used to forecast how
323	future air quality regulations will affect surface ozone, they must simulate ozone within
324	the correct production regime (i.e., NO <sub>x</sub> -limited vs. VOC-limited). For the Baltimore
325	area, this updated model platform increases the percentage of the ozone formed in a NO <sub>x</sub> -
326	limited regime from ~75 to ~85% of the total. Since the updated model platform places
327	ozone in a more NO <sub>x</sub> -limited regime, it is possible a simulation of surface ozone long-
328	term trends using these changes will resolve the long-standing difficulty in simulating the
329	response of surface ozone to past reductions in ozone precursors.

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- Anderson, D. C., C. P. Loughner, G. Diskin, A. Weinheimer, T. P. Canty, R. J. Salawitch,
- 343 H. M. Worden, A. Fried, T. Mikoviny, A. Wisthaler, and R. R. Dickerson (2014),
- Measured and modeled CO and NOy in DISCOVER-AQ: An evaluation of emissions
- and chemistry over the eastern US. *Atmos. Environ.*, 96, 78-87.
- 346 Appel, K. W., A. B. Gilliland, G. Sarwar, and R. C. Gilliam (2007), Evaluation of the
- Community Multiscale Air Quality (CMAQ) model version 4.5: Sensitivities impacting
- model performance Part I Ozone. Atmos. Environ., 41(40), 9603-9615.
- 349 doi:10.1016/j.atmosenv.2007.08.044
- 350 Appel, K. W., C. Chemel, S. J. Roselle, X. V. Francis, R. Hu, R. S. Sokhi, S.T. Rao, and
- S. Galmarini (2012), Examination of the Community Multiscale Air Quality (CMAQ)
- model performance over the North American and European domains. *Atmos. Environ.*, 53, 142-155.
- Bash, J.O., K.R. Baker, and M.R. Beaver (2015), Evaluation of improved land use and
- canopy representation in BEIS v3.61 with biogenic VOC measurements in California.
- 356 *Geosci. Model Dev. Disc.*, *8*, 8117–8154.
- Bey, I., D. J. Jacob, R. M. Yantosca, J. A. Logan, B. D. Field, A. M. Fiore, Q. B. Li, H.
- G. Y. Liu, L. J. Mickley, and M. G. Schultz (2001), Global modeling of tropospheric
- 359 chemistry with assimilated meteorology: Model description and evaluation. *J. Geophys.* 360 *Res.-Atmos.*, 106(D19), 23073-23095.
- Brioude, J., W. M. Angevine, R. Ahmadov, S.-W. Kim, S. Evan, S. A. McKeen, E.-Y.
- Hsie, G. J. Frost, J. A. Neuman, I. B. Pollack, J. Peischl, T. B. Ryerson, J. Holloway, S.
- S. Brown, J. B. Nowak, J. M. Roberts, S. C. Wofsy, G. W. Santoni, T. Oda, and M.
- 364 Trainer (2013), Top-down estimate of surface flux in the Los Angeles Basin using a
- mesoscale inverse modeling technique: assessing anthropogenic emissions of CO, NO<sub>x</sub>,
- and CO2 and their impacts. Atmos. Chem. Phys., 13, 3661-3677.
- Canty, T. P., L. Hembeck, T. P. Vinciguerra, D. C. Anderson, D. L. Goldberg, S. F.
- Carpenter, D. J. Allen, C. P. Loughner, R. J. Salawitch, and R. R. Dickerson (2015),
- Ozone and NO x chemistry in the eastern US: evaluation of CMAQ/CB05 with satellite (OMI) data. *Atmos. Chem. Phys.*, *15*, 10965-10982.
- Carlton, A. G, and K. R. Baker (2011), Photochemical Modeling of the Ozark Isoprene
- Volcano: MEGAN, BEIS, and Their Impacts on Air Quality Predictions. *Environ. Sci.*
- 373 *Technol.*, 45, 4438-4445.
- Castellanos, P., L. T. Marufu, B. G. Doddridge, B. F. Taubman, J. J. Schwab, J. C. Hains,
- S. H. Ehrman, and R. R. Dickerson (2011), Ozone, oxides of nitrogen, and carbon
- monoxide during pollution events over the eastern United States: An evaluation of
- emissions and vertical mixing. J. Geophys. Res.-Atmos., 116, D16307.
- 378 doi:10.1029/2010JD014540
- Chameides, W. L., F. Fehsenfeld, M. O. Rodgers, C. Cardelino, J. Martinez, D. Parrish,
- W. Lonneman, D. R. Lawson, R. A. Rasmussen, P. Zimmerman, J. Greenberg, P.
- Middleton, and T. Wang (1992), Ozone precursor relationships in the ambient
- atmosphere. Journal of Geophys. Res.-Atmos., 97(D5), 6037-6055
- Crawford, J. H., R. R. Dickerson, and J. Hains (2014), DISCOVER-AQ Observations and
- and early results. *Environmental Manager*, 8-15.

- 385 Dolwick, P., F. Akhtar, K. R. Baker, N. Possiel, H. Simon, and G. Tonnesen (2015)
- 386 Comparison of background ozone estimates over the western United States based on
- two separate model methodologies. Atmos. Environ., 109, 282–296. 387
- 388 Doraiswamy, P., C. Hogrefe, W. Hao, R. F. Henry, K. Civerolo, J. Ku, G. Sistla, J. J.
- Schwab, and K. L. Demerjian (2009), A diagnostic comparison of measured and 389
- 390 model-predicted speciated VOC concentrations. Atmos. Environ., 43, 5759-5770.
- 391 Emery, C., J. Jung, N. Downey, J. Johnson, M. Jimenez, G. Yarwood, and R. Morris
- 392 (2012), Regional and global modeling estimates of policy relevant background ozone 393 over the United States. Atmos. Environ., 47, 206–217.
- 394 Emmons, L. K., S. Walters, P. G. Hess, J.-F. Lamarque, G. G. Pfister, D. Fillmore, C.
- 395 Granier, A. Guenther, D. Kinnison, and T. Laepple (2010), Description and evaluation
- 396 of the Model for Ozone and Related chemical Tracers, version 4 (MOZART-4). Geosci. 397 Model Dev., 3(1), 43-67.
- 398 EPA. (2014a). Modeling Guidance for Demonstrating Attainment of Air Quality Goals 399 for Ozone, PM2.5, and Regional Haze. Retrieved November 2015 from
- 400
- http://www.epa.gov/ttn/scram/guidance/guide/Draft O3-PM-RH Modeling Guidance-401 2014.pdf
- EPA. (2014b). Meteorology Technical Support Document Meteorological Model 402
- Performance for Annual 2011 WRF v3.4 Simulation. Retrieved November 2015 from 403
- 404 http://www.epa.gov/ttn/scram/reports/MET TSD 2011 final 11-26-14.pdf
- EPA. (2014c). Technical Support Document (TSD) Preparation of Emissions Inventories 405
- 406 for the Version 6.2, 2011 Emissions Modeling Platform. Retrieved November 2015
- 407
- http://www3.epa.gov/ttn/chief/emch/2011v6/2011v6 2 2017 2025 EmisMod TSD a 408 ug2015.pdf 409
- 410 EPA. (2014d). Air Quality Designations for the 2008 Ozone National Ambient Air
- Quality Standards. Retrieved November 2015 from 411
- http://www.epa.gov/oagps001/greenbk/hindex.html 412
- 413 Fehsenfeld, F. C., R. R. Dickerson, G. Hubler, W. T. Luke, L. J. Nunnermacker, E. J.
- Williams, J. M. Roberts, J. G. Calvert, C. M. Curran, and A. C. Delany (1987), A 414
- ground-based intercomparison of NO, NOx, and NOy measurement techniques. 415
- 416 Journal of Geophys. Res.-Atmos., 92(12), 710–722.
- Ferreira, J., A. Rodriguez, A. Monteiro, A. Miranda I, M. Dios, J. A. Souto, G. Yarwood, 417
- U. Nopmongcol, and C. Borrego (2011), Air quality simulations for North America-418
- 419 MM5-CAMx modelling performance for main gaseous pollutants. Atmos. Environ., 52, 420 212-224
- 421 Fiore, A. M., J. T. Oberman, M. Y. Lin, L. Zhang, O. E. Clifton, D. J. Jacob, V. Naik, L.
- 422 W. Horowitz, J. P. Pinto, and G. P. Milly (2014), Estimating North American
- background ozone in U.S. surface air with two independent global models: Variability, 423
- uncertainties, and recommendations. Atmos. Environ., 96, 284-300. 424
- doi:10.1016/j.atmosenv.2014.07.045 425
- Foley, K. M, C. Hogrefe, G. Pouliot, N. Possiel, S. J. Roselle, H. Simon, and B. Timin 426
- (2015), Dynamic evaluation of CMAQ part I: Separating the effects of changing 427
- emissions and changing meteorology on ozone levels between 2002 and 2005 in the 428
- 429 eastern US. Atmos. Environ., 103, 247-255.

- 430 Fujita, E. M., D. E. Campbell, B. Zielinska, J. C. Chow, C. E. Lindhjem, A. DenBleyker,
- 431 G. A. Bishop, B. G. Schuchmann, D. H. Stedman, and D. R. Lawson (2012),
- Comparison of the MOVES2010a, MOBILE6.2, and EMFAC2007 mobile source 432
- 433 emission models with on-road traffic tunnel and remote sensing measurements. Journal
- of the Air & Waste Management Association 62, no. 10: 1134-1149. 434
- 435 Gilliland, A. B., C. Hogrefe, R. W. Pinder, J. M. Godowitch, K. L. Foley, and S. T. Rao
- 436 (2008), Dynamic evaluation of regional air quality models: assessing changes in O 3
- 437 stemming from changes in emissions and meteorology. Atmos. Environ., 42, 5110-438 5123.
- 439 Goldberg, D. L., T. P. Vinciguerra, K. M. Hosley, C. P. Loughner, T. P. Canty, R. J.
- Salawitch, and R. R. Dickerson (2015), Evidence for an increase in the ozone 440
- 441 photochemical lifetime in the eastern United States using a regional air quality model.
- Journal of Geophys. Res.-Atmos., 120, 12,778–12,793. 442
- 443 Goldberg, D. L., C. P. Loughner, M. Tzortziou, J. W. Stehr, K. E. Pickering L. T.
- 444 Marufu, and R. R. Dickerson (2014), Higher surface ozone concentrations over the
- 445 Chesapeake Bay than over the adjacent land: Observations and models from the
- 446 DISCOVER-AQ and CBODAQ campaigns. Atmos. Environ., 84, 9-19.
- Guenther, A. B., X. Jiang, C. L. Heald, T. Sakulyanontvittaya, T. Duhl, L. K. Emmons, 447
- and X. Wang (2012), The Model of Emissions of Gases and Aerosols from Nature 448
- 449 version 2.1 (MEGAN2.1): an extended and updated framework for modeling biogenic
- 450 emissions. Geosci. Model Dev., 5(6), 1471-1492. doi:10.5194/gmd-5-1471-2012
- He, H., J. W. Stehr, J. C. Hains, D. J. Krask, B. G. Doddridge, K. Y. Vinnikov, T. P. 451
- 452 Canty, K. M. Hosley, R. J. Salawitch, and R. R. Dickerson (2013), Trends in emissions
- and concentrations of air pollutants in the lower troposphere in the 453
- Baltimore/Washington airshed from 1997 to 2011. Atmos. Chem. Phys., 13, 1-16. 454
- Henderson, B. H., F. Akhtar, H. O. T. Pye, S. L. Napelenok, and W. T. Hutzell (2014), A 455
- database and tool for boundary conditions for regional air quality modeling: description 456 457 and evaluation. Geosci. Model Dev., 7(1), 339-360.
- 458 Hildebrandt-Ruiz, L., and G. Yarwood (2013), Interactions between Organic Aerosol and
- NOy: Influence on Oxidant Production. Final report for AQRP project 12-012. 459
- Prepared for the Texas Air Quality Research Program. 460
- 461 Hogrefe, C., J. Biswas, B. Lynn, K. Civerolo, J. Y. Ku, J. Rosenthal, C. Rosenzweig, R.
- Goldberg, and P. L. Kinney (2004), Simulating regional-scale ozone climatology over 462
- the eastern United States: model evaluation results. Atmos. Environ., 38, 2627-2638. 463
- Horowitz, L. W., A. M. Fiore, G. P. Milly, R. C. Cohen, A. Perring, P. J. Wooldridge, P. 464
- G. Hess, L. K. Emmons, and J. F. Lamarque (2007), Observational constraints on the 465
- chemistry of isoprene nitrates over the eastern United States. J Geophys Res-Atmos, 466
- 112, D12S08. doi:10.1029/2006JD007747 467
- Houyoux, M. R., and J. M. Vukovich (1999), Updates to the Sparse Matrix Operator 468
- Kernel Emissions (SMOKE) modeling system and integration with Models-3. The 469
- Emission Inventory: Regional Strategies for the Future, 1461. 470
- Kinosian, J. R. (1982), Ozone precursor relationships from EKMA diagrams. *Environ*. 471
- 472 Sci. Technol. 16, no. 12: 880-883.
- Koo, B., N. Kumar, E. Knipping, U. Nopmongcol, T. Sakulvanontvittava, M. T. Odman, 473
- A. G. Russell, and G. Yarwood (2015), Chemical transport model consistency in 474

- simulating regulatory outcomes and the relationship to model performance. *Atmos*.
- 476 Environ., 116, 159–171.
- Kota, S. H., G. Schade, M. Estes, D. Boyer, and Q. Ying (2015), Evaluation of MEGAN
- predicted biogenic isoprene emissions at urban locations in Southeast Texas. *Atmos. Environ.* 110, 54-64.
- 4/9 Environ. 110, 34-04.
- Loughner, C. P., D. J. Allen, K. E. Pickering, D. L. Zhang, Y. X. Shou, and R. R.
- Dickerson (2011), Impact of fair-weather cumulus clouds and the Chesapeake Bay
- breeze on pollutant transport and transformation. *Atmos. Environ.*, 45(24), 4060–4072.
- 483 Mao, J., F. Paulot, D. J. Jacob, R. C. Cohen, J. D. Crounse, P. O. Wennberg, C. A. Keller,
- 484 R. C. Hudman, M. P. Barkley, and L. W. Horowitz (2013), Ozone and organic nitrates
- over the eastern United States: Sensitivity to isoprene chemistry. *J. Geophys. Res.- Atmos.*, *118(19)*, 11,256–11,268.
- 487 Martins, D. K., R. M. Stauffer, A. M. Thompson, T. N. Knepp, and M. Pippin (2012),
- Surface ozone at a coastal suburban site in 2009 and 2010: relationships to chemical
- and meteorological rocesses. *Journal of Geophys. Res.-Atmos.* 117(D5), D05306.
- Perring, A. E., S. E. Pusede, and R. C. Cohen (2013), An observational perspective on the
- atmospheric impacts of alkyl and multifunctional nitrates on ozone and secondary organic aerosol. *Chemical Reviews*, 113, 5848 5870.
- Pouliot, G., and T. E. Pierce (2009), Integration of the Model of Emissions of Gases and
- 494 *Aerosols from Nature (MEGAN) into the CMAQ Modeling System*. Retrieved from http://www3.epa.gov/ttnchie1/conference/ei18/session3/pouliot.pdf
- Ramboll Environ. (2014). *CAMx Version 6.10 User's Guide*. Retrieved November 2015 from http://www.camx.com/files/camxusersguide v6-10.pdf
- Sillman, S. (1999), The relation between ozone, NOx and hydrocarbons in urban and polluted rural environments. *Atmos. Environ.*, *33*(12), 1821-1846.
- 500 Simon, H., K. R. Baker, and S. Phillips (2012) Compilation and interpretation of
- 501 photochemical model performance statistics published between 2006 and 2012. *Atmos. Environ.* 61, 124-139.
- Skamarock, W. C., J. B. Klemp, J. Dudhia, D. O. Gill, D. M. Barker, W. Wang, and J. G.
- Powers. 2008, A description of the advanced WRF version 3. *NCAR technical note NCAR/TN/u2013475+ STR*.
- Thompson, A. M., R. M. Stauffer, S. K. Miller, D. K. Martins, E. Joseph, A. J.
- Weinheimer, and G. S. Diskin (2014), Ozone profiles in the Baltimore-Washington
- region (2006 2011): satellite comparisons and DISCOVER-AQ observations. *J. Atmos. Chem.* 1-30.
- Warneke, C., J. A. de Gouw, L. Del Negro, J. Brioude, S. McKeen, H. Stark, W. C.
- Kuster, P. D. Goldan, M. Trainer, F. C. Fehsenfeld, C. Wiedinmyer, A. B. Guenther, A.
- Hansel, A. Wisthaler, E. Atlas, J. S. Holloway, T. B. Ryerson, J. Peischl, L. G. Huey,
- and A. T. Case Hanks (2010), Biogenic emission measurement and inventories
- determination of biogenic emissions in the eastern United States and Texas and
- comparison with biogenic emission inventories. J. Geophys. Res.-Atmos. 115, D00F18
- Williams, E. J., K. Baumann, J. M. Roberts, S. B. Bertman, R. B. Norton, F. C.
- Fehsenfeld, S. R. Springston, L. J. Nunnermacker, L. Newman, and K. Olszyna (1998)
- Intercomparison of ground based NOv measurement techniques. *Journal of Geophys.*
- 519 Res.-Atmos., 103(D17), 22261–22280.

Yarwood, G, S. Rao, M. Yocke, and G. Whitten (2005), Updates to the Carbon Bond 520 chemical mechanism: CB05. Final report to the US EPA, RT-0400675, 8. 521 Yu, S., R. Mathur, J. Pleim, G. Pouliot, D. Wong, B. Eder, K. Schere, R. Gilliam, and ST 522 523 Rao. 2012. Comparative evaluation of the impact of WRF/NMM and WRF/ARW meteorology on CMAQ simulations for PM2. 5 and its related precursors during the 524 2006 TexAQS/GoMACCS study. Atmos. Chem. Phys 12, 4091-4106. 525 526 Zhou, W., D. S. Cohan, and S. L. Napelenok. 2013. Reconciling NO<sub>x</sub> emissions 527 reductions and ozone trends in the US, 2002-2006. Atmos. Environ. 70, 236-244.

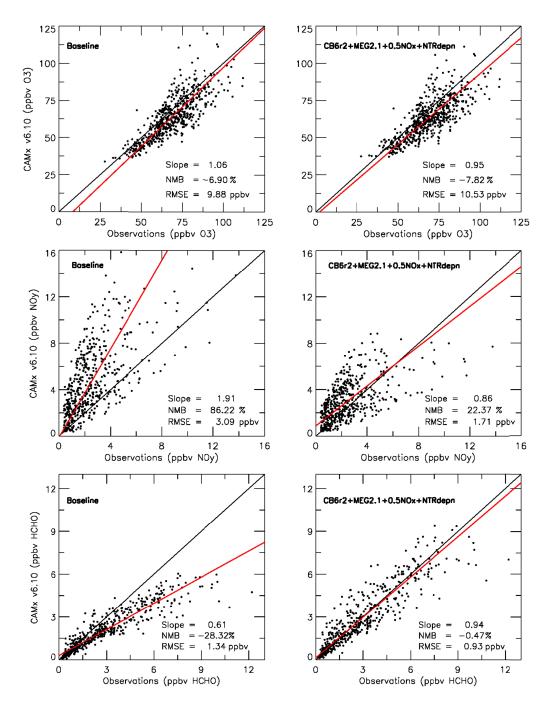


Figure 1. Observations acquired by the P3-B aircraft during DISCOVER-AQ Maryland in July 2011 compared to model output from CAMx v6.10 at the nearest model grid point and closest hourly interval. The closest hourly model output is matched to each one-minute averaged P3-B observation; both quantities are then averaged over the same tenminute interval. Left panels show the baseline simulation, while right panels show the updated "Beta" simulation. Top row shows O<sub>3</sub>, middle row shows NO<sub>y</sub>, and bottom row shows HCHO. Black lines represent the 1:1 line, while red lines represent the linear best fit.

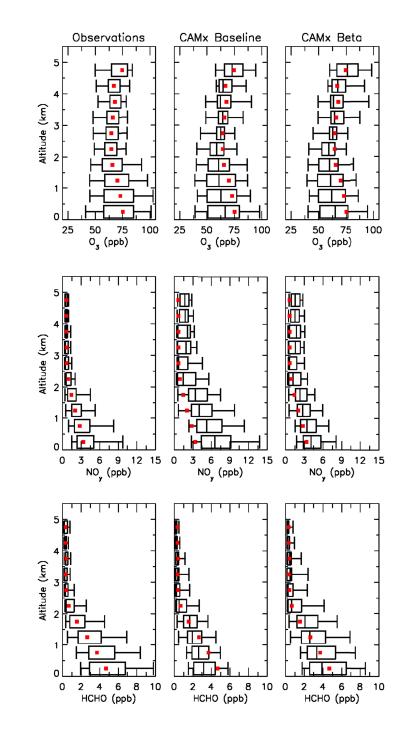
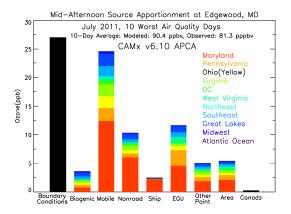


Figure 2. Vertical profiles of  $O_3$ ,  $NO_y$ , and HCHO binned in 500 m intervals, showing the  $5^{th}$ ,  $25^{th}$ ,  $50^{th}$ ,  $75^{th}$  and  $95^{th}$  percentiles. Left panels show one-minute averaged data from the P3-B aircraft, center panels show the baseline simulation, and the right panels show the updated "Beta" simulation. Model output from CAMx v6.10 is matched spatially and temporally to the P3-B measurements at one-minute intervals. Top row shows  $O_3$ , middle row shows  $NO_y$ , and bottom row shows HCHO. Red squares indicate the median values of the observations, which are shown on all panels to facilitate visual comparison.



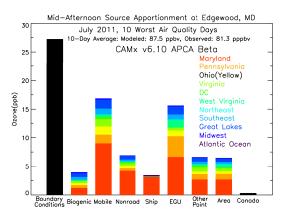


Figure 3. Ozone attributed to source sectors separated by U.S. states and the region of Canada that is in the modeling domain (Figure S1) during the ten worst air quality days in July 2011 at 2 PM local time at the Edgewood, MD monitoring site, located 30 km east-northeast of Baltimore: (left) baseline simulation and (right) updated "Beta" simulation.

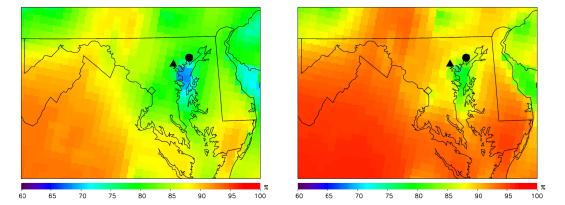


Figure 4. Percentage of ozone formed in the  $NO_x$ -limited production regime during July 2011 averaged over daytime (8 AM – 8 PM local time) for the entire month in the (left) baseline simulation and (right) updated "Beta" simulation. The filled triangle denotes Baltimore, Maryland and the filled circle denotes Edgewood, Maryland.

